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Biomass- An Overview on Composition Characteristics and Properties

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ABSTRACT

A general overview has been proposed for biomass composition, characteristics and property. Biomass is the combination of all organic materials that simply referring to amount of animal and plant. Biomass materials are diverse, ranging from wood, bark, straw and other agricultural residues, grasses and off-spec grains. Despite of diversity the composition of most biomass materials is relatively uniform, when moisture has been excluded. Biomass burning emission is an important part of carbon and nitrogen cycle. The pollutants released by biomass burning varies depends upon the constituents of nitrogen sulphur and carbon. In this paper basic knowledge of biomass and comparative data regarding composition and characterization has been studied.

Keywords: Biomass, Categorization, Composition, Property.

Introduction

Biomass consists of bio + mass that simply referring to amount of animal and plant. It refers to the organic matter deriving from plants and that is generated through photosynthesis.

Biomass is combination of all organic material that stems from plants (including algae, plant, trees and crops). It is produced by green plants converting sunlight into plant material through photosynthesis and includes all land and water based vegetation, as well as all organic wastes (Kendry 2002). When burned, the energy stored in biomass is released to produce heat or electricity. Common forms of solid biomass include agricultural crops, crop residue and forestry products (Jenkins., et.al.1998).

For Plant material the carbon used to construct biomass is absorbed from the atmosphere as carbon dioxide (CO₂) by plant life, using energy from the sun. Plants may subsequently be eaten by animals and thus converted into animal biomass. However the primary absorption is performed by plants.

If plant material is not eaten it is generally either broken down by micro-organisms or burned: If broken down it releases the carbon back to the atmosphere, mainly as either carbon dioxide (CO₂) or methane (CH₄), depending upon the conditions and processes involved. If burned the carbon is returned to the atmosphere as CO₂.

These processes have happened for as long as there have been plants on Earth and is part of what is known as the carbon cycle.

From the perspective of energy resources, a common definition is “a general term for animal and plant resources and the wastes arising from them, which have accumulated in a certain amount (excluding fossil resources).”

- Biomass is an abundant and renewable source of energy.
- Using biomass for energy would diversify the energy supply and reduce dependency on fossil fuels.

In this paper i have tried to discuss about basic knowledge of biomass, its origination, availability, categories of bioresources and property of biomass. From the entire study also discussed about the ultimate analysis of biomass. This paper will be helpful in our next study related to study of emission from biomass burning.

Origination of biomass

Carbon dioxide (CO₂) from the atmosphere and water absorbed by the plant roots are combined in the photosynthetic process to produce carbohydrates (or sugars) that form the biomass. Or Biomass is the

plant material derived from the reaction between Carbon dioxide (CO₂) in the air, water and sunlight, via photosynthesis to produce carbohydrate that forms the building blocks of biomass (Kendry 2002).

The solar energy that drives photosynthesis is stored in the chemical bonds of the biomass structural components. During biomass combustion, oxygen from the atmosphere combines with the carbon in biomass to produce CO₂ and water. The process is therefore cyclic because the carbon dioxide is then available to produce new biomass. This is also the reason why bio-energy is potentially considered as carbon-neutral, although some CO₂ emissions occur due to the use of fossil fuels during the production and transport of biofuels (Broek et.al., 1996).

Availability of bio resources

The earth has a huge stock biomass covering wide regions including forests and the ocean. The total biomass of the world is 1,800 billion tons on the ground and 4 billion tons in the ocean, and a comparative amount of biomass exists in the soil. The total biomass on the ground is 33,000 EJ on the energy basis, which corresponds to 80 times or more of the annual energy consumption of the world (Stegmann, 1996).

However some part of the biomass is used as food by living things including humans, and also for uses other than foods which are necessary for the human living.

Categories of Biomass resources

There is no established way of categorizing biomass, which is defined differently according to the field; categorization changes depending on the purpose and application. Generally there are two ways to categorize biomass: one is biological categorization based on types of existing biomass in nature and the other is based on the use or application as resources.

An example of biomass categorization appears in figure 1(Yokoyama, 2008). In this categorization, biomass includes not only the conventional product and waste from agriculture, forestry and fisheries, but also plantation biomass.

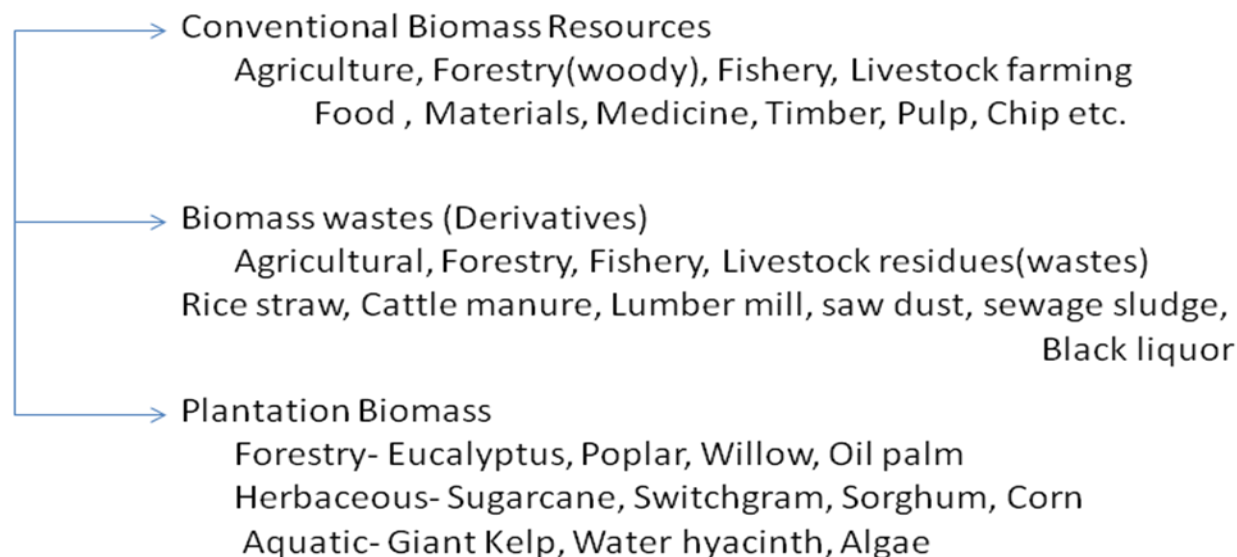


Fig.1. Biomass Categorization (in terms of use and application), *Source:* Yokoyama, 2008;

According to the different sectors biomass resources can be classified differently has been shown in Table 1(www.ieabcc.nl).

Table 1. Biomass classification on the basis of different sector, *Source:* (www.ieabcc.nl).

Sectors	Type	Examples
Forestry	Dedicated forestry	Short rotation plantations (e.g. willow, poplar, eucalyptus)
	Forestry by-products	Wood blocks, wood chips from thinnings
Agriculture	Dry lignocellulosic energy crops	Herbaceous crops (e.g. miscanthus, reed canarygrass, giant reed)
	Oil, sugar and starch energy crops	Oil seeds for methylesters (e.g. rape seed, sunflower)
		Sugar crops for ethanol (e.g. sugar cane, sweet sorghum)
		Starch crops for ethanol (e.g. maize, wheat)
	Agricultural residues	Straw, prunings from vineyards and fruit trees
Livestock waste	Wet and dry manure	
Industry	Industrial residues	Industrial waste wood, sawdust from sawmills
		Fibrous vegetable waste from paper industries
Waste	Dry lignocellulosic	Residues from parks and gardens (e.g. prunings, grass)
	Contaminated waste	Demolition wood
		Organic fraction of municipal solid waste
		Biodegradable landfilled waste, landfill gas
		Sewage sludge

Biomass fuels consist of three main segments: wood, waste and alcohol fuels and the hierarchy of biomass energy resources has been shown in figure 2 (Energy Information Administration/Estimates of U.S.Biomass Energy Consumption 1992).

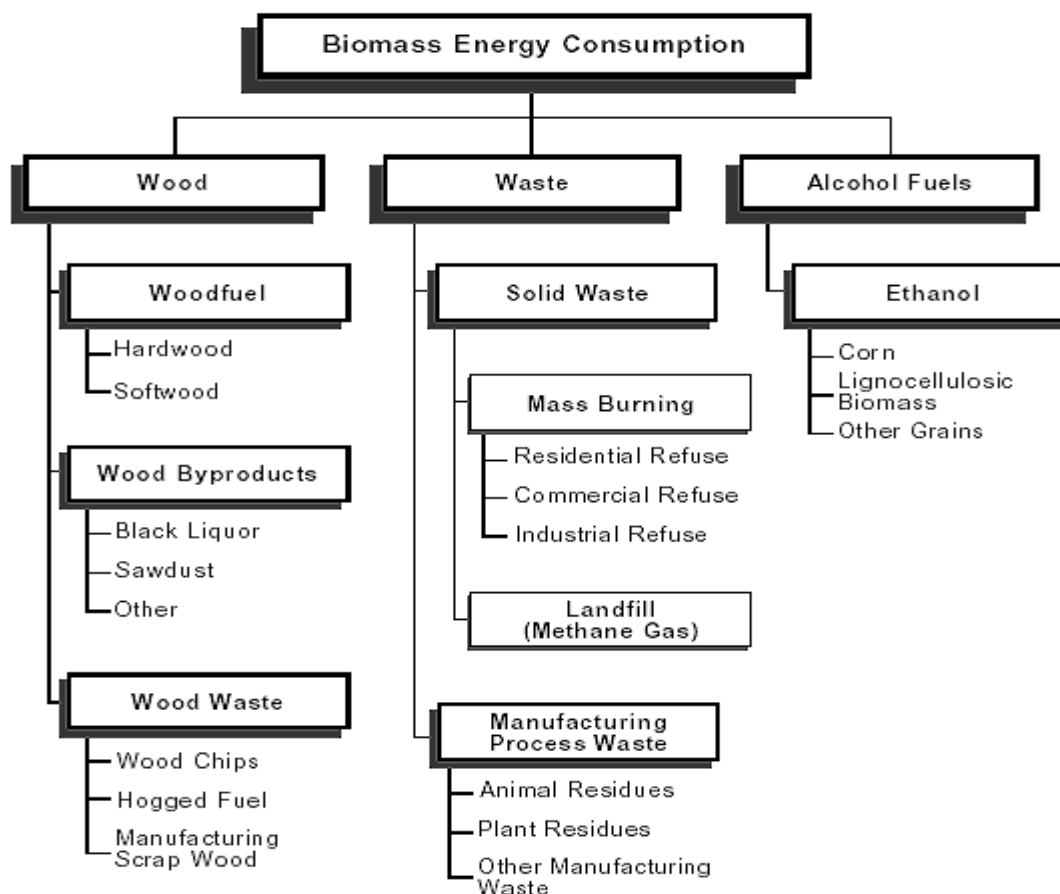


Figure 2. Biomass Energy Resource Hierarchy, *Source:* Energy Information Administration/Estimates of U.S. Biomass Energy Consumption 1992.

Biomass composition

There is a wide variety of biomass, and composition varies significantly. Some primary components are cellulose, hemicelluloses, lignin, starch and proteins. Trees mainly consist of cellulose, hemicelluloses and lignin(Yokoyama, 2008).

- a) Cellulose- A polysaccharide in which D-glucose is linked uniformly by β -glucosidic bonds. Its molecular formulae is $(C_6H_{12}O_6)_n$. The degree of polymerization, indicated by n, is broad, ranging from several thousand to several ten thousands.
- b) Hemi cellulose- A polysaccharide whose units are 5-carbon monosaccharide including D-xylose and D-arabinose and 6-carbon monosaccharide's including D-mannose, D-galactose and D-glucose.
- c) Lignin- A compound whose constituents units, phenylpropane and its derivatives are bonded 3-dimensionally, cellulose, hemicelluloses and lignin are universally

found in many kinds of biomass and are the most plentiful natural carbon resources on earth.

The chemical composition of plant biomass varies among species. In general terms, plants are made of approximately 25% lignin and 75% carbohydrates or sugars. The carbohydrate fraction consists of many sugar molecules linked together in long chains or polymers. Two categories are distinguished: cellulose and hemi-cellulose. The lignin fraction consists of non-sugar type molecules that act as a glue holding together the cellulose fibers. Table 2(www.ecn.nl/phyllis) shows typical values of cellulose, hemi-cellulose and lignin for the composition of straw, softwoods and hardwoods.

Table 2. Typical values for the composition of straw, softwoods and hardwoods, *Source:* (www.ecn.nl/phyllis).

	Cellulose(%)	Hemi-cellulose(%)	Lignin(%)
Softwood	45	25	30
Hardwood	42	38	20
Straw stalks	40	45	15

Energy content of biomass

The heating value of a fuel indicates the energy available in the fuel per unit mass-MJ/kg(BTU/lb). The net heating value is the actual energy available for heat transfer. The difference in available energy is explained by the fuel’s chemical composition, moisture and ash content.

Moisture

Moisture content is the key factor determining the net energy content of biomass material. The most important property of biomass feedstocks with regard to combustion-and to the other thermo-chemical processes-is the moisture content, which influences the energy content of the fuel. Dry biomass has a greater heating value, as it uses little of its energy to evaporate any moisture. There is a correlation between energy and moisture contents. Increased moisture means less energy available. Table 3(www.vt.tuwien.ac.at/biobib) shows possible ranges in moisture content for selected biomass resources and table 4 shows the evolution of the lower heating value (LHV, in GJ/t) of wood as a function of the moisture content.

All biomass materials contain some moisture, from as low as 10% for dried straw to over 50% for fresh-cut wood.

Table 3. Possible ranges in moisture content for selected biomass resources, *Source:* www.vt.tuwien.ac.at/biobib

Biomass resources	Moisture content (%)
Industrial fresh wood chips and sawdust	40-60 wt. % (wb)
Industrial dry wood chips and sawdust	10-20 wt. % (wb)
Fresh forest wood chips	40-60 wt. % (wb)
Chips from wood stored and air-dried several months	30-40 wt. % (wb)
Waste wood	10-30 wt. % (wb)
Dry straw	15 wt. % (wb)

Table 4. Some typical characteristics of biomass fuels compared to oil and coal,

Typical characteristics Fuel	GJ/t	toe/t	Kg/m ³	GJ/ m ³	Volume oil equivalent(m ³)
Fuel oil	41.9	1,00	950	39,8	1,0
Coal	25.0	0,60	1000	25,0	1,6
Pellets 8% moist.	17.5	0,42	650	11,4	3,5
Pile wood (stacked, 50%)	9.5	0,23	600	5,7	7,0
Industrial softwood chips 50% moist.	9.5	0,23	320	3,0	13,1
Industrial softwood chips 20% moist.	15.2	0,36	210	3,2	12,5
Forest softwood chips 30% moist.	13.3	0,32	250	3,3	12,0
Forest hardwood chips 30% moist.	13.3	0,32	320	4,3	9,3
Straw chopped 15% moist.	14.5	0,35	60	0,9	45,9
Straw big bales 15% moist.	14.5	0,35	140	2,0	19,7

Source: www.vt.tuwien.ac.at/biobib

Ash

The non-combustible content of biomass is referred to as ash. High ash content leads to fouling problems, especially if the ash is high in metal halides (e.g., potassium). Unfortunately, biomass fuels, especially agricultural crops/residues tend to have a high ash with high potassium content. Wood (core, no bark) has less than 1% ash. Bark can have up to 3% ash. Agricultural crops have higher ash content, from 3% and higher (Figure 3).

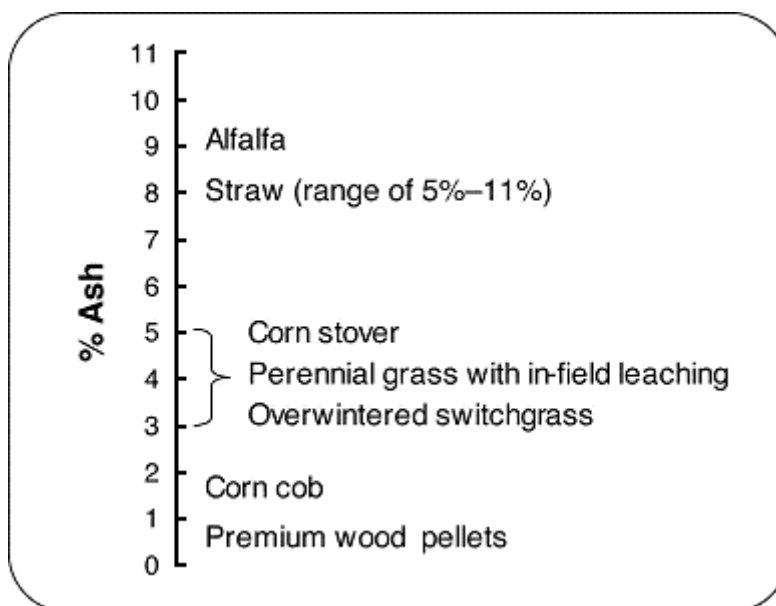


Figure 3. Typical ash content for selected biomass on a dry basis. Source: AURI, 2005; Preto, 2010.

Carbon

The carbon content of biomass is around 45%, while coal contains 60% or greater (Demirbas, 2004). A higher carbon content leads to a higher heating value.

Hydrogen

The hydrogen content of biomass is around 6% (Jenkins, 1998). A higher hydrogen content leads to a higher heating value.

Nitrogen

The nitrogen content of biomass varies from 0.2% to more than 1% (Jenkins, 1998). Fuel-bound nitrogen is responsible for most nitrogen oxide (NOx) emissions produced from biomass combustion. Lower nitrogen content in the fuel should lead to lower NOx emissions.

Sulphur

Most biomass fuels have a sulphur content below 0.2%, with a few exceptions as high as 0.5%–0.7%. Coals range from 0.5%–7.5% (Demirbas 2004). Sulphur oxides (SOx) are formed during combustion and contribute significantly to particulate matter (PM) pollution and acid rain. Since biomass has negligible sulphur content, its combustion does not contribute significantly to sulphur emissions.

Chloride

Combustion of biomass with high chloride concentrations (over 1,000 µg/g) can lead to increased ash fouling. High chloride content leads to the formation of hydrochloric acid in the boiler tubes, resulting in corrosion that can lead to tube failure and water leaks in the boiler. Fuels where this has been observed include corn stover and corn cobs.

Properties of Biomass

The ultimate analyses for a variety of biomass materials are presented in Table 5(Auri,2005; Preto, 2010). All results are displayed on a dry matter basis for comparison. By this table we can compare the % values of ash, carbon, hydrogen, nitrogen, sulphur, oxygen and total chlorine for different biomass.

Table 5. Ultimate analysis for a variety of biomass fuels in Ontario (all values reported on a dry matter basis), Source: AURI, 2005; Preto, 2010.

Biomass Type	MJ/kg	BTU/lb	Typical Values ¹						
			Ash %	Carbon %	Hydrogen %	Nitrogen %	Sulphur %	Oxygen % ²	Total Chlorine (µg/g) ³
Off-spec (non-food) grains									
Beans	19	7,996	4.7	45.7	6.3	4.3	0.7	38.8	193
Corn	17	7,350	1.5	42.1	6.5	1.2	0.1	48.9	472
Canola	28	12,220	4.5	60.8	8.3	4.5	0.5	21.4	163
Dried distillers grain	22	9,450	4.9	50.4	6.7	4.7	0.7	32.6	1,367
Grass/forages									
Big blue stem	19	8,020	6.1	44.4	6.1	0.8	0.1	42.6	1,880
Miscanthus	19	8,250	2.7	47.9	5.8	0.5	0.1	43.0	1,048
Sorghum	17	7,240	6.6	45.8	5.3	1.0	0.1	42.3	760
Switchgrass	18	7,929	5.7	45.5	6.1	0.9	0.1	41.7	1,980
Straw/residue									
Alfalfa	17	7,435	9.1	45.9	5.2	2.5	0.2	39.5	3,129
Barley straw	17	7,480	5.9	46.9	5.3	0.7	0.1	41.0	1,040
Corn cobs	18	7,927	1.5	48.1	6.0	0.4	0.1	44.0	2,907

Corn stover	19	7,960	5.1	43.7	6.1	0.5	0.1	44.6	1,380
Flax straw	18	7,810	3.7	48.2	5.6	0.9	0.1	41.6	2,594
Wheat straw	18	7,710	7.7	43.4	6.0	0.8	0.1	44.5	525
Processing by-product									
Oat hulls	19	7,960	5.1	46.7	6.1	0.9	0.1	41.1	1,065
Soybean hulls	18	7,720	4.3	43.2	6.2	1.8	0.2	44.3	266
Sunflower hulls	20	8,530	4.0	47.5	6.2	1.0	0.2	41.2	3,034
Wood									
Bark	19	8,432	1.5	47.8	5.9	0.4	0.1	45.4	257
Willow	19	8,550	2.1	50.1	5.8	0.5	0.1	41.4	134
Hardwood	19	8,300	0.4	48.3	6.0	0.2	0.0	45.1	472
Coal									
Low sulphur subbit coal – PRB ⁴	25	10,520	6.0	55.0	3.7	0.9	0.4	11.5	35
Lignite	22	9,350	22.0	58.8	4.2	0.9	0.5	13.6	25

¹ The content level of ash, chlorine and other elements can be lowered through crop selectivity, growing conditions, plant fractionation, harvest time and harvest method.

² Calculated by difference. Percent by difference refers to the difference between two numbers as a percent of one of them. For example, the percentage difference from 5 to 3 is: $2/5 = 0.4 = 40\%$.

³ A microgram (μg) is a unit of mass equal to $1/1,000,000$ of a gram (1×10^{-6}), or $1/1,000$ of a milligram. It is one of the smallest units of mass commonly used.

⁴ PRB – Power River Basin

It is important to note that biomass materials naturally contain variability, which depends on:

- geographical location
- variety
- climate conditions
- harvest methods

Summary

Biomass is combination of all organic materials. The process of biomass derivation to exhaustion is part of carbon cycle. Biomass is an abundant and renewable source of energy. The origination of biomass is derived from the reaction between carbon dioxide via photosynthesis. On energy basis the total biomass available on ground corresponds to 80 times or more of the annual consumption of the world. Biomass is categorized by different means like depending on use, on the basis of different sectors and on the basis of energy resources.

For most biomass fuels, nitrogen and sulphur levels are quite low, that mean resulting in relatively low SO_x and NO_x emissions. This type of biomass will have less pollutant generation tendency. The major difference in composition of biomass fuel is ash content. The ash content is maximum for grasses and agricultural residues above 5% for bark 2-3% and for traditional biomass fuel wood contains less than 0.5% ash. Maximum ash content cause significant foulig problems like grasses and agricultural residues. On the basis of this study we will try to give our overview in respect of biomass burning emission and its

relation with the composition of biomass and different characteristics like ash content, moisture content etc.

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References

1. Kendry, Mc P., 2002. Energy production from biomass (part 1): overview of biomass. *Bioresource Technology*. 83, 37-46.
2. Jenkins, B.M., Baxter, L.L., Miles, T.R., 1998. Combustion properties of biomass. *Fuel Processing Technology*. 54, 17-46.
3. Broek, V., Faaij, R.A., Wijk, A., 1996. Biomass combustion for power generation. *Biomass and Bioenergy*. 11(4), 270-281.
4. Stegmann, R., 1996. Landfill gas utilization: An overview in landfilling of waste: Biogas (ed T.H.Christensen, R.Cossu, R.Stegmann), E&FN SPON.
5. Yokoyama, S., 2008, *The Asian Biomass Handbook*. A guide for biomass production & utilization. The Japan Institute of Energy.
6. AURI. 2005. *Agricultural renewable solid fuels data*. Retrieved from Agricultural Utilization Research Institute Fuels Initiative website.
7. Preto, F., 2010. Properties of the 13 common biomass fuels in Ontario. Natural Resource Canada (NRCan), Ottawa, ON.
8. Demirbas, A., 2004. Combustion characteristics of biomass fuels. *Progress Energy Combustion Science*. 30, 219-230.
9. Energy information administration/Estimates of U.S.Biomass Energy Consumption 1992.
10. www.ieabcc.nl, www.ecn.nl/phyllis, www.vt.tuwien.ac.at/biobib