

A Review on Environmentally-safe Corrosion deterrence of Metals in Various Media

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DOI: <http://dx.doi.org/10.21013/jas.v4.n1.p6>

How to cite this paper:

Jothi, R., Saratha, R., & Priya, S. (2016). A Review on Environmentally-safe
Corrosion deterrence of Metals in Various Media. *IRA-International Journal of Applied
Sciences* (ISSN 2455-4499), 4(1). doi:<http://dx.doi.org/10.21013/jas.v4.n1.p6>

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ABSTRACT

The present review gives a brief application of environmentally – safe corrosion inhibitors for various metals like aluminum, copper, mild steel, and zinc. Due to the enormous increase of industrial activities, a large amount of mineral acids are consumed. Therefore more number of metals gets corroded. These problems can be solved by using corrosion deterrence. Although many chemical deterrence are available most of them are highly virulent. So there is a need to explore a new class of corrosion inhibitors with good inhibition efficiency and that are environmentally - safe. Consequently, it can be achieved by using plant materials as corrosion inhibitors which are non-toxic, easily available and biodegradable.

Key words: Aluminum, Mild steel, copper, Zinc, Eco-friendly corrosion inhibitors, plant extract

Introduction:

Recently, a lot of investigations have been carried out in search of different types of corrosion inhibitors for various types of metals, in different media. Although chemical inhibitors (organic & inorganic) are available, they cause much damage to our surroundings. Therefore to overcome these problems, the area of research regarding the plant extracts as corrosion inhibitors is being concentrated much because of their renewability, low-cost, eco-friendliness and easy availability. The extract of leaves, stem, bark, root, fruit and seed consists of mixture of organic compounds and has the ability to function as effective inhibitors for metal in acid, alkaline and neutral media. The inhibition process occurs via the phytoconstituents present in the plant material on the surface of the metal, thus a protective film is formed therein corrosion process is prevented further.

A list of plant extracts as eco-friendly corrosion inhibitors is given below:

S.no	Metal/Medium	Inhibitor/ Corrosion parameters		Phytochemicals present	Methods adopted	Major findings	Ref no
1.	Aluminum alloy AA3003/ 0.4 M HCl and 0.5 M HCl	<i>Aspilia africana</i> leaf		Vitamin B	Gravimetric, Electrochemical, Quantum chemical computations & Molecular simulations	Both Physisorption and Chemisorption process occurred.	1
		Conc	100-700 mg/L				
		Temp	303-333 K				
2.	Aluminum/ 1 M HCl	Coconut coir dust		Lignins,Tannins,Cellulose, Pentosan,Furfural	Weight loss and Hydrogen evolution methods	Chemisorption IE - 80% at 60°C –Weight loss method	2
		Conc	0.1 - 0.5 g/L				
		Temp	30°C and 60°C				
3.	Aluminum/ 2M HCl	<i>Chlomolaena odorata</i> L. leaf		Oils,Steroids, Triterpenes, Flavonoids	Gasometric and Thermometric techniques	Physisorption, IE - 92.16% (Gasometric) IE - 95.12%(Thermometric) at 7.5 % V/V	3
		Conc	10-50 % V/V				
		Temp	30°C and 60°C				
4.	Aluminum/ 0.5 mol/L HCl	<i>Cocos nucifera</i> L.		Amino acids, Ascorbic acid, Vitamin B1,Sorbitol,Fructose, Glucose,Malic acid, Phytohormone,Auxin,Cytokinin, Gibberellin,	Weight loss Method	Spontaneous adsorption IE - 93%	4
		Conc	0 - 7.5 % V/V				
		Temp	30 °C				
5.	Aluminum alloy (7075)/ 3.5% NaCl	Date palm (<i>Phoenix dactylifera</i> L.) Fruit juice		Cellobiose,Glucose,Xylose, Fructose,Arabinose,	Potentiodynamic polarization and EIS	Physisorption Cathodic inhibitor	5
		Conc	100 - 2000 ppm				
		Temp	--				
6.	Aluminium and 6063 aluminium alloy/1.0 M H ₃ PO ₄	<i>Garcinia indica</i>		Garcinol,Curcumin	Tafel Extrapolation and Electrochemical impedance techniques,	Mixed-type inhibitor, Chemisorption, IE of <i>Garcinia indica</i> extract was efficient for 6063 alloy when compared with Al (99.6%)	6
		Conc	100 - 500ppm				
		Temp	30-50°C				
7.	Mild steel /1 M H ₂ SO ₄	<i>Araucaria Columnaris</i> (Gum Exudates)		Polysaccharide(1,2- benzenedicarboxylic acid, bis(2-ethylhexyl) ester, diisooctyl- phthalate, phthalic acid, isobutyl and isopropyl ester)	Weight loss, Electrochemical method, SEM	Maximum inhibition efficiency is achieved at very lower concentration (400 ppm), Mixed type inhibitor	7
		Conc	50 - 600 ppm				
		Temp	303 - 323 K				

8.	Aluminium and AA5754 aluminium alloy/ 3% NaCl	<i>Laurus nobilis</i> L.oil		1,8-cineole, Methyl eugenol, α -terpinyl acetate, Linalool, Sabinene	Weight loss method, Potentiodynamic polarization and Linear Polarization method, SEM	AA5754 alloy has better corrosion resistance in 3% NaCl solution than pure alloy	8
		Conc	0 – 50 ppm				
		Temp	25 °C				
9.	Aluminum alloy (Al7075)/ 3.5% NaCl	L-Glutamine		Amino acid	Polarization and Electrochemical impedance studies	The critical electrode rotation rate for the best performance of inhibitor is 1000 rpm	9
		Conc	0-2000 rpm				
		Temp	--				
10	Aluminium/0.5 mol/L HCl. KCl, KBr are added as additives	<i>Morinda tinctoria</i> leaves		Alkaloids, Flavonoids, Terpenoids, Steroids, Amino acids	Mass loss and electrochemical techniques, SEM	IE - 96.72% With immersion of 2 h. Langmuir adsorption. Physisorption	10
		Conc	0.5-7.0%				
		Temp	303-333k				
11	Aluminum & Aluminum silicon alloy/0.5 M HCl	Phoenix dactylifera Leaves		Gallic acid, Protocatechuic acid, Ferulic acid, β -carotene, Lutein, Quercetin, Luteolin	Potentiodynamic Polarization, EIS & EFM	Mixed type Cathodic inhibitor, Physisorption	11
		Conc	200-1000 ppm				
		Temp	20°C				
12	Mild steel /1 M HCl	Galinsoga parviflora (Quick Weed)		Alkaloids, Flavanoids, Saponins, Tannins, Phytosterol, Galinsosides	Weight loss, Potentiodynamic polarization and EIS	IE - 90% at 6 % V/V, Mixed-type inhibitor	12
		Conc	2.5 – 6 % V/V				
		Temp	303-343 K				
13	Mild steel /1 M HCl	Bombax malabaricum seeds		Arachidic acid, Cyclopropane, Lignoceric acid, Gossypol, Linoleic acid, Myristic acid, Palmitic acid, Oleic acid and Stearic acid	Weight loss, Potentiodynamic polarization and EIS	IE – 96.47% at 2.5 % V/V , Mixed-type inhibitor,	13
		Conc	0.005%, 0.05%, 0.15%, 0.5%, 2.5% (V/V)				
		Temp	303-343 K				
14	Copper/0.5 M Sulfuric acid	<i>Myrtus Communis</i> (oil extract)		Limonene, Linalool, Pinene, 1,8-Cineole, Linanyl Acetate, And Terpeneol,	Weight Loss, Potentiodynamic Polarization, EIS, SEM & AFM	Mixed-type inhibitor affecting both the Cathodic and anodic reactions	14
		Conc	25-100mg/L				
		Temp	-				
15	Copper/2.5 M HNO ₃	Vitamin B1 hydrochloride (Thiamine Hydrochloride)		--	Weight loss study	IE-69.9% at 2mM thiamine hydrochloride	15
		Conc	0.01-2.5 mM				
		Temp	30,35,40&45 \pm 0.2 °C				

16	AISI steel 4130/1M HCl	Taurine (2-aminoethanesulfonic acid)		--	Potentiodynamic Polarization, EIS, AFM & Optical microscopy	Mixed-type inhibitors, Chemisorption	16
		Conc	12.5-250 ppm				
		Temp	25°C, 45 °C, 65 °C				
17	Carbon steel (ASTM 1015)/ 1.0 mol L ⁻¹ HCl	Antibacterial drugs (Penicillin G, Ampicillin & Amoxicillin Drugs)		--	Potentiodynamic polarization, EIS and Electrochemical noise (EN)	Mixed-type inhibitor	17
		Conc	0.1 to 10.0 mM				
		Temp	25-55 °C				
18	C38 Steel/ 1 M HCl	Black pepper extract and its piperine		Piperine, Piperanine	Weight loss method	IE - 95.8 % at 2 g/L	18
		Conc	0.008 – 2 g/L				
		Temp	40 - 70 °C				
19	C38 steel/0.5 M H ₂ SO ₄	<i>Lavandula multifida</i> L		Carvacrol	Weight loss, Electrochemical Polarization & EIS methods	Mixed-type inhibitor, IE-72.2% at 2 g/L of oil at 298 K	19
		Conc	0.25-2 g/L				
		Temp	303-343 K				
20	Mild steel/ 1 M HCl	<i>Tagetes erecta</i> stem		β -caryophyllene (8.5%), terpinolene (18.4%), (E)-ocimene (12.6%), (Z) β -ocimene (10.4%), piperitenone (10.4%), (Z)-ocimene (5.5%), limonene (6.2%)	Chemical and Electrochemical studies, FTIR	IE - 97% at 0.3% V/V, Mixed- type inhibitor	20
		Conc	0.001 -0.3 % V/V				
		Temp	30°C				
21	Mild Steel/ 0.5 M, 1 M, 2 M HCl and H ₂ SO ₄	Acacia trees (Exudate gum)		Polysaccharides like 42% Galactosyl, 27% Arabinosyl, 15% Rhamnosyl, 14.5% Glucuronosyl 1.5% 4-O-methyl-glucuronosyl and Protein- polysaccharide (minor component)	Weight loss, Hydrogen evolution, Electrochemical polarization, SEM, FTIR, XPS	Mixed-type inhibitor, Inhibition efficiency increases in the presence of magnetic field	21
		Conc	0.6 mg/L				
		Temp	30°C				
22	Mild steel/ 1 M HCl	<i>Alstonia anugustifolia</i> Var. <i>latifolia</i> (A. <i>latifolia</i>) Leaves		Indole alkaloids (Alstogustine 19-epialstogustine Alstopirocine)	EIS, SEM, FTIR	Mixed- type inhibitor, IE-84% for Impedance studies, IE-88% for Polarization studies	22
		Conc	1-5 mg/L				
		Temp	30±2 °C				

23.	Mild steel /0.5 M H ₂ SO ₄	<i>Anacyclus pyrethrum</i> L. (leaves and stem, flowers, roots)		Catechol tannins, saponins, alkaloids, flavonoids (absent in roots), amino acids	EIS, Potentiodynamic Polarization	IE-87.01% for leaves & stems, 88.88% for flowers, and 79.30% for roots, Chemisorption	23
		Conc	100-350 mg/L				
		Temp	25- 40 °C				
24	C38 steel/ 1 M HCl	<i>Aniba rosaeodora</i>		Anibine	Potentiodynamic polarization, EIS, XPS, NMR	Mixed –type inhibitor, Anibine as the major constituent(EIS)	24
		Conc	25-200 mg/L				
		Temp	25 °C				
25	Mild steel /HCl	<i>Areca catechu</i> (Seed extract)		polyphenols (flavonols and tannins), alkaloids (Arecoline, Arecaidine, Guvacoline, and Guvacine), carbohydrates, fats, proteins, crude fiber, and mineral matters	Weight loss, polarization & impedance studies, FTIR, SEM	Mixed mode of inhibition	25
		Conc	0-500 ppm				
		Temp	303-318 K				
26	Mild steel/ 4 N HCl & Conc. HCl	<i>Artemisia pallens</i>		Tetrahydrofuran, Davanone, Linalool, Ethylcinnamate, Sesquiterpenoids, Terpenoids, Artemone	Weight loss, Polarization - 4 N HCl, Weight-loss, SEM & FTIR-Conc. HCl	IE-93% at 1.5 g/L in 4 N HCl IE-96.5% at 40 g/L in Conc. HCl	26
		Conc	0.1-1.5 g/l				
		Temp	30 ± 1 °C				
27	Mild steel/1 M HCl	<i>Aquilaria crassna</i> Leaves extracts crude extract(ME) and Aqueous Extract(AE)		Saponins(higher presence), Flavanoids & Tannins(moderate presence), Alkaloids & Triterpenoids(Week presence)	Weight loss, Potentiodynamic polarization, EIS, XPS SEM, FTIR	Max IE at 300 ppm for ME 80.28% (Weight loss), 66%(EIS) , 67% (Potentiodynamic Polarization) and Max IE at 60 ppm for AE 69.98 % (weight loss), 60% (Potentiodynamic Polarization) and 59%(EIS) at 50 ppm	27
		Conc (ME)	0-300 ppm				
		Conc (AE)	0-60 ppm				
		Temp	27 °C				
28	Mild steel/1 N HCl	Basella alba L. leaves		Alkaloids, Carbohydrates, Pseudo tannins, Chlorogenic acids,	Weight loss, Potentiodynamic	Maximum inhibition efficiency was	28
		Conc	65 – 95 ppm				

		Temp	-	Steroidal glycosides, Coumarin and Flavonoids	polarization, EIS, SEM	81.28% at the inhibitor concentration 95 ppm	
29	Mild steel/ 1 M HCl	<i>Caulerpa racemosa</i>		Sesquiterpenes Caulerpin an alkaloid Reducing sugars Galactose,glucose, arabinose xylose, mannose, rhamnose,amino acids	Weight loss, EIS,UV,IR,NMR & AFM	Mixed-mode of inhibition IE- 83%	29
		Conc	0-25 ppm				
		Temp	303-323 K				
30	Mild steel/ 0.5 M H ₂ SO ₄	<i>Centella asiatica</i> (leaves)		Triterpene acids (Asiatic, Terminolic, Centic, Centillic Centoic Acid) Alkaloids(hydrocotylin), volatile fatty oils(glycerites of palmitic,stearic, lignoceric, oleic,lynoleic and linolenic acids), glycosides(asiaticoside A, asiaticoside B, madecassoside, centellic acid),and flavonoids(3- glucosylquercetin, 3-glucosylkaempferol and 7- glucosylkaempferol	Gravimetric, EIS,SEM	IE-95.08% for 1200 ppm at 303 K, Mixed-type inhibitor	30
		Conc	0-1200 ppm				
		Temp	30 - 60 °C				
31	Mild steel/1 M HCl & 0.5 M H ₂ SO ₄	<i>Chlorophytum borivilianum</i> root		Saponins	Weight loss, EIS, SEM,FTIR,UV- Visible spectroscopy	Mixed –type inhibitor	31
		Conc	300 - 500 mg/L				
		Temp	35 – 55 °C				
32	Mild steel/ 1 M HCl	<i>Clematis gouriana</i> Leaves		Magnoflorine, Aporphine alkaloid,	Weight loss, Potentiodynamic polarization, EIS,FTIR SEM,EDS	IE-95.70%. Possibility of both physisorption and chemisorption	32
		Conc	0 – 400 ppm				
		Temp	300 - 330 K				
33	Mild steel/1 M HCl & H ₂ SO ₄	<i>Ervatamia coronaria</i> (leaf extract)		Ervatamine-An alkaloid,indole Alkaloids,Tri-terpenoids	Weight loss,EIS SEM, XRD	Mixed mode inhibitor	33
		Conc	10-50 ppm				
		Temp	303-323K				
34	Mild steel/ 2 M HCl	Fig leaves		-	Weight loss Measurement,	IE-87% for concentration above 200 ppm,	34
		Conc	0 – 1000 ppm				

		Temp	25±2°C		Electrochemical techniques	Physisorption,	
35	Mild steel/ 2.5 M H ₂ SO ₄	Gnetum Africana (Ethanol extract)		Alkaloid,Saponin, Tannin,Terpene, Anthraquinone,Flavonoid, Cardiac glycoside, Phlobatanins, Amino acids	Weight loss, Gasometric, Thermometric methods, FTIR	Physisorption	35
		Conc	0.1 0.5 g/L				
		Temp	303 – 333 K				
36	Mild steel/1M HCl & H ₂ SO ₄	<i>Hydroclathrus clathratus</i> (marine alga)		N-[2-hydroxy-1- (hydroxymethyl)pentadecyl]eicosan amide, Uracil, 1-(2-deoxy-b-D-ribofuranosyl)-5- methyl-1,2,3,4- tetrahydropyrimidine-2,4-dione, and 2-phenylamino-1,4-naphthoquinone	Weight loss, EIS, SEM & AFM	Mixed-mode of inhibition, IE-65.28% for 1M HCl IE-77.64% for 1 M H ₂ SO ₄ (Impedance studies) IE-76.6% for 1M HCl IE-85.02% for 1M H ₂ SO ₄ (Polarization studies) at 500 ppm	36
		Conc	100 - 500 ppm				
		Temp	303 - 323K				
37	Mild steel/1 M H ₂ SO ₄ and 1 M HCl	<i>Kigelia pinnata</i> Leaves		Saponins, Tannins, Phenolic acids, Flavonoids, Alkaloids, and Carbohydrates	Weight loss,Tafel polarization,EIS, X-ray diffraction, SEM,FTIR,UV- Visible Spectroscopy	The inhibitor inhibits the corrosion reaction in both acid environments and inhibition efficiency follows the order H ₂ SO ₄ > HCl	37
		Conc	25 -125 ppm				
		Temp	308 K,313 K,318 K,323 K & 328 K				
38	Mild steel/ 1 M HCl and 0.5 M H ₂ SO ₄	<i>Syzygium cumini</i> seed		Ellagic acid, Gallic acid, Quercetin, Cafeic acid	Weight loss, EIS,	The inhibition efficiency is slightly better in HCl solution, Mixed –type inhibitor	38
		Conc	60 - 600 ppm				
		Temp	308–338 K				
39	6063 Aluminium alloy / 1 M phosphoric acid	<i>Coriandrum sativum</i> L. seed extract		Linalool (Major constituent)	Potentiodynamic polarization, EIS,SEM	Mixed inhibitor,Results obtained by Tafel polarisation method and EIS method were in good agreement with one another. SEM studies confirmed the corrosion inhibition of the alloy due to the adsorption of the inhibitor	39
		Conc	100 - 500 ppm				
		Temp	30°C to 50°C				
40	Carbon steel/ 1 M HCl	<i>Sesbania sesban</i>		--	Weight loss and Potentiodynamic Polarization techniques	IE - 91.08%, Mixed type inhibitor	40
		Conc	0.2 - 2 g/l				
		Temp	25°C to 65°C				

Abbreviations:

NMR: Nuclear Magnetic Resonance Spectroscopy
 EIS: Electrochemical Impedance Spectroscopy
 AFM: Atomic Force Microscopy
 SEM: Scanning Electron Microscopy
 FTIR: Fourier Transform Infrared Spectroscopy
 EDS: Energy Dispersive Spectrum
 EFM: Electrochemical Frequency Modulation

Conclusion:

From the review article, it can be seen that the common phytoconstituents responsible for corrosion inhibition are saponins, alkaloids, flavonoids, and phenolic compounds. The major techniques adopted to evaluate the corrosion inhibition efficiency are : Weight –loss method, Electrochemical impedance studies, Thermometric method, Gasometric method. The surface morphology techniques used are scanning electron microscope (SEM), Energy - dispersive X-ray analysis (EDAX) and Fourier transform infra-red spectroscopy which confirms the formation of protective film on the corroded metal therein to stop the corrosion process.

All the plant materials have been evaluated as corrosion inhibitors in the laboratory level, unless it is being used in industries the research is not of much use. Hence transfer of technology is essential. These plant materials as corrosion inhibitors may be tested in various industries, where chemical inhibitors are already in use. To apply the plant materials as inhibitor in the industries the inhibitors should be standardized for which the reliability and reproducibility tests should be carried out. Hence more research could be carried out using plant material as corrosion inhibitors in the applied field.

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